



Thermal conductivity enhancement of phase change materials : A literature review

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Received 21 Sept 2019,
Revised 03 June 2020,
Accepted 04 June 2020

Keywords

- ✓ PCMs,
- ✓ Thermal conductivity,
- ✓ Carbon material,
- ✓ Metal foams,

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Abstract

Recently the higher demand of fossil fuels and increasing attention of risks related to environmental issues have contributed to find other renewable sources of energy use to the different industrial and daily domestic activities. Due to this the use of phase change materials (PCMs) in thermal energy storage systems attract more and more attention in recent years due to their contribution to enhanced energy efficiency. However, their low thermal conductivity presents one of the issues that should be resolved. The necessity of increasing the thermal conductivity of PCMs is evident due to its low charging/discharging rates which is important during thermal cycling. In order to overcome the issue of low thermal conductivity. Several studies have been developing new type of PCMs by introducing high thermal conductive filler such as carbon additives or metal foams. Carbon materials are so far the best fillers to increase the effective thermal conductivity of PCM since only small volume fraction is needed and they have high thermal conductivity and low density. The thermal conductivity enhancement using carbon material is the point of focus of this work. Many studies have been done on improving the heat transfer of PCMs. In general, the studies are divided into several categories depending on the types of thermal conductivity enhancers.

1. Introduction

Due to the higher energy demand of fossil fuels and climate change awareness the use of different renewable energy sources is attracting a serious attention [1-3]. The industrial, commercial and domestic activities; the electrical energy consumption varies considerably between peak and off peak periods. Better power generation management can be achieved if some of the peak load could be transferred to the off peak load period, which can be achieved by the thermal energy storage which is widely available in nature in form of solar and geothermal energy.

As a matter of fact, thermal energy can be stored as latent energy by heating and cooling a material. Phase change materials are mostly used to store such thermal energy [4-10]. However, their low thermal conductivity is one of the main problems that should be overcome [11-16]. Higher thermal conductivity could be achieved by using some additives with the aim of increasing the energy charging/discharging rates. Additive could be classified into material based carbon or metal foams. The thermal conductivity enhancement is the point of focus of this review.

2. Thermal conductivity enhancement of phase change materials

2.1. Thermal conductivity enhancement of phase change materials using carbon material

2.1.1. Expanded graphite

Xu et al. [15] prepared $\text{LiNO}_3 - \text{KCl} - \text{NaNO}_3/\text{EG}$ composite PCM using the capillary method with 5 wt%, 10 wt%, 15 wt%, 20 wt%, and 25 wt% EG. At 25wt% mass fraction of EG, they found out that no leakage was observed. Using the SEM images, it has been proven that $\text{LiNO}_3 - \text{KCl} - \text{NaNO}_3$ composites were intensely absorbed by EG. In their experiments, the thermal conductivity and the compress density show a strong linear relationship. However, the thermal conductivity of the eutectic salt under a compress density of $1.482\text{g}/\text{cm}^3$ was $1.608\text{W}/\text{mK}$. When the compress density of the composite PCM varies from 0.9 to $2\text{g}/\text{cm}^3$; the thermal conductivity varies from 18.57 to $31.53\text{W}/\text{mK}$ which was 11.5-19.6 higher than that of the pure salt. Zichen et al. [16] use also EG as a high thermal conductivity enhancer and $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$ as PCM. It has low thermal conductivity, liquid leakage, high supercooling and poor thermal reliability after numerous thermal cycles. To overcome those issue, EG with different mass fraction (9 wt%, 13 wt%, 16.67 wt%, and 20 wt%) was mixed with $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$. The thermal conductivity data of composite PCMs with 9 wt%, 13 wt%, 16.67 wt%, and 20 wt% of EG was respectively measured to be $0.942\text{W}/\text{mK}$, $1.053\text{W}/\text{mK}$, $1.354\text{W}/\text{mK}$ and $1.658\text{W}/\text{mK}$. Also, DSC analysis showed that the addition of EG decreased the degree of supercooling by 29.4°C . Zhong et al [17] prepared three kinds of porous composite phase change materials using three types of binary salts ($\text{LiNO}_3 - \text{KCl}$, $\text{LiNO}_3 - \text{NaNO}_3$ and $\text{LiNO}_3 - \text{NaCl}$) as the phase change materials and expanded graphite as the high thermal conductive additive. The thermal conductivity was significantly increased. It is analyzed that, after impregnation of the salt into the EG, the thermal conductivity of $\text{LiNO}_3 - \text{KCl}$ is intensified by 5.0 times; $\text{LiNO}_3 - \text{NaNO}_3$, 6.9 times; and $\text{LiNO}_3 - \text{NaCl}$, 4.9 times. Yuping et al. [18] developed a novel shape stabilized phase change material by impregnating hydrated salts into expanded graphite. He found out that the thermal conductivity of hydrated salts/EG was high as $3.615\text{W}/\text{mK}$ (uncoated) and $3.646\text{W}/\text{mK}$ (coated with paraffin wax). Heking et al. [19] prepared $\text{NaCl}-\text{CaCl}_2/\text{EG}$ composite phase change materials at different fraction of EG. It is found that the thermal conductivity increased linearly with the content of EG and the density of the composite. At 20 wt% of EG and $2\text{g}/\text{cm}^3$, the thermal conductivity was $4.937\text{W}/\text{mK}$ higher which 701.1 % higher than binary eutectic chloride. Huang et al. [20] prepared a $\text{LiNO}_3/\text{KCl} - \text{EG}$ composite phase change material. Increasing the content of EG in the composite from 10% to 30%, the thermal conductivity has been improved from 1.85 times to 6.65 times respectively. Jumbing et al. [21], prepared $\text{NaNO}_3-\text{KNO}_3/\text{EG}$ with different mass rates of EG. The results showed that the addition of EG have give a great rise to thermal conductivity coefficient which has increased to $4.884\text{W}/\text{mK}$.

2.1.2. Carbon fiber

Recently, carbon materials have become one of the best fillers for enhancing the thermal conductivity, due to the fact that they possess high thermal conductivity and low density. There have been many studies on the preparation of phase change material composites using carbon fiber [22-25]. Zhao et al. [26] used graphite fiber to prepare Carbon bonded carbon fiber (CBCF) monoliths. The CBCF were filled with paraffin wax to for the PCM composite. The in-plane thermal conductivity of the PCM composite was significantly increased up to 57 times over the pure wax, while the out-of-plane thermal conductivity was also increased by 3.7 to 5.5 times. In addition, the improvements in thermal conductivity showed almost linear relationship with the volume fraction of carbon fibers in the PCM composites. Takahiro et al. [27] prepared a phase change material composite with a percolating network of Carbon Fiber. Two types of methods have been used: The conventional melting dispersion method and a novel hot press method. Erythritol was chosen as the phase change material and carbon fiber as the high thermal conductive filler. It is found that the Phase material composite prepared by the hot press method presents a high thermal conductivity; furthermore, It is found that the Phase material composite prepared by the hot press method presents a high thermal conductivity, furthermore, a percolating filler network can be easily formed in this case as it is shown in Fig.1. Frusteri et al. [28] investigated the influence of carbon fiber on the thermal conductivity enhancement of an organic PCM44. Different

lengths of carbon fiber have been randomly distributed and the thermal conductivity has been measured. It was demonstrated that a linear relationship exists between the carbon fiber content and the thermal conductivity. The best results were obtained using a micro-fiber (length=0.2mm).

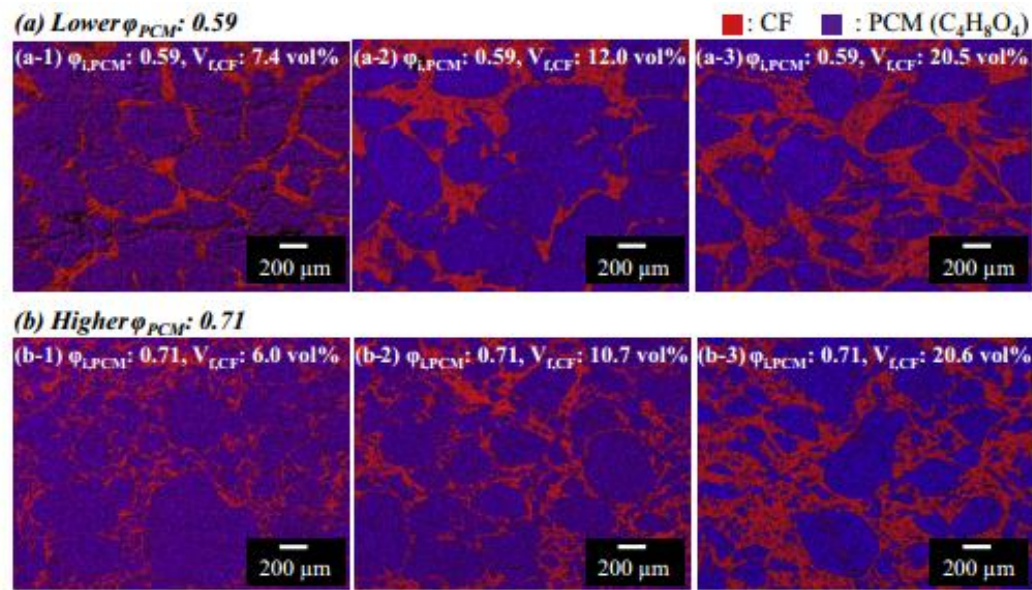


Figure 1: EDS mapping from the cross section of the PCCs: (a) a lower packing ratio of 0.59 and (b) a higher packing ratio of 0.71 [27]

2.1.3. Carbon nanotubes

Tingting et al. [29] reported the preparation of a new polyethylene glycol (PEG)/diatomite form-stable phase change composite (fs-PCC) with single-walled carbon nanotubes (SWCNTs) as nano-additive. The obtained results showed that the thermal conductivity of the PCC has been increased from 0.24 W/mK in case of pure PEG to 0.87 W/mK with a small fraction of SWCNTs (2wt%). Zhang et al. [30] studied the effect of the introduction of CNTs as an additive for improving the thermal conductivity of paraffin-carbon nanotubes/expanded perlite form-stable composite phase change materials (PA-CNTs/EP FS-CPCMs). The vacuum impregnation method was used to prepare the composite with different mass fractions of CNT. The thermal conductivity of PA-CNTs/EP FS-CPCMs 5.27 (0.516 Wm⁻¹ K⁻¹) was 4.82 times that of PA-CNTs/EP FS-CPCMs 0. The thermal storage and release properties of PA-CNTs/EP FS-CPCMs were significantly improved as compared with those of PACNTs/EP FS-CPCMs 0. Feng et al. [31] prepared a new composite where the MWCNTs were chosen as the high thermal conductivity enhancer, Na₂CO₃ was used as phase change material and MgO as a supporting material. With the increase of MWCNTs amount and temperature of use, the thermal conductivity has been significantly increased. Feng et al. [32] reported the preparation of Na₂CO₃/MgO composite phase change materials with added Multi-walled carbon nanotubes (MWCNTs). Na₂CO₃ was used as phase change material and MgO as a supporting material. It is found that the thermal conductivity increased with an increase of the MWCNTs loading and an increase in the temperature of use.

In order to improve the performance of salt phase change materials, Tao et al. [33] prepared carbonate salt/nanomaterial composite phase change material using four kinds of carbon nanomaterials with different microstructures as Fig. 2 illustrates. The results show that nanomaterial with columnar structure such as single-walled carbon nanotubes (SWCNT) are more efficient at enhancing the thermal conductivity of the PCM salt up to 56.98%.

2.1.4. Comparison of different types of carbon additives:

The comparison of the thermal conductivity enhancement using the different types of carbon additives are listed in Table 1. The following conclusions can be drawn:

- The thermal conductivity is significantly increased in case of high aspect ratio of carbon additive

- The addition of small amount of CNTs can contribute to the larger enhancement of thermal conductivity
- The EG and graphite powder are promising additives to enhance the thermal conductivity

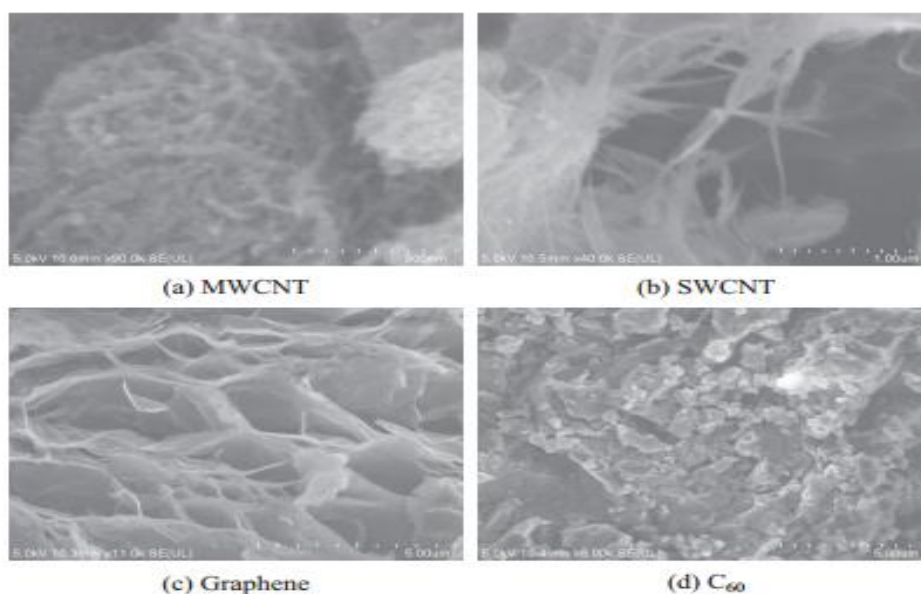


Figure 2: Carbon nanomaterial microstructure by SEM. [33]

Table 1: Comparison of the thermal conductivity enhancement with some types of carbon additives

Carbon additive	PCM	k_{PCM} (W/m.K)	Fraction	k_{PCC} (W/m.K)	Increase (%)
Expanded Graphite	Eutectic LiNO ₃ /KCl [20]	1.749	30 wt%	11.63	665
	Ternary eutectic chloride [34]	1.174	5 wt%	2.084	178
	Form stable PCM [35]	0.43	20 wt%	4.59	1067
Carbon fiber	Erythritol [27]	0.733	25 vol%	≈30	4000
Grafted CNTs	Paraffin [36]	0.2312	4 wt%	0.7903	342
Short SWCNT	Water [37]	0.580	0.48 vol%	0.604	104
Long SWCNT	Water [37]	0.580	0.48 vol%	0.627	108
MWCNT	Water [37]	0.580	0.48 vol%	0.598	103
Graphite powder	Bakelite [38]	1.4	30 vol%	4.84	346
Graphene	Paraffin [39]	0.25	10 wt%	2.7	1080

Table 2: Comparison of the thermal conductivity enhancement with some types of metal additives

Metal	PCM	k_{PCM} (W/m.K)	Fraction	k_{PCC} (W/m.K)	Increase (%)
Copper foam[40]	Paraffin	0.354	88.89 %	16.01	4500
Nickel foam [40]	Paraffin	0.354	90.61 %	2.33	658
β -AlN powder [41]	Polyethylene glycol	0.2985	30 wt%	0.7661	257%
Al [42]	Paraffin	0.25	9 wt%	5	2000%

2.2. Thermal conductivity enhancement of phase change materials using metal foam

Recently, more interest was given to the development of high thermal conductive PCMs using metal foams. High thermal conductivity, porosity and strong mixing capability are the main properties of metal foams which make it as one of the most promising high thermal conducting additives. [Table 2](#) lists some of the experimental results.

Xiao et al [43] prepared paraffin/metal foam composite PCM where copper and nickel were the metallic foam matrix. The results show that the thermal conductivity of the paraffin increased three times more than the pure paraffin. Thapa et al [44] used the copper foam as the high thermal conductive filler and icosane wax as the PCM for low thermal energy storage application. The thermal conductivity of the composite was 3.78 W/mK which is higher than that of the pure icosane wax.

Bauer and Wirtz [45] developed a plate like structural thermal energy storage composite consisting of a central core of aluminum foam packed with PCM. The effective thermal conductivity of the composite is significantly increased. Jiang et al. [46] prepared two kinds of shape-stabilized PCMs composites using Al foam impregnated with paraffin or stearic acid and the results show that the PCMs were well impregnated into the Al foam because with no crack appearing on the surface of the PCMs.

Li et al [47] prepared a salt hydrate/copper foam composite phase change material using Sodium acetate trihydrate (SAT) as thermal energy storage phase change material. The thermal conductivity of copper foam/SAT composite PCM is about 11 times higher than that of pure SAT. Moreover, the charging and discharging times can be reduced significantly. Additionally, the charging time is only 40% of that of the pure SAT. Huang et al [48] fabricated Myristyl alcohol (MA)/metal foam composite phase change materials (CPCMs) by vacuum melting infiltration of nickel foam and copper foam; MA acts as the PCM. Effects of pore size and thermal properties were analyzed and compared. There was reported inverse relation between pore diameters. The thermal conductivity of the MA/nickel foam and MA/copper foam was increased by 1.80 and 7.51 times respectively as compared to pure MA.

Metal salts are another form of metallic additives with high thermal conductivity and reliable stability. Fauzi et al. [49] found that adding 10% sodium laurate to Myristic acid (MA)/palmitic acid (PA) eutectic mixture leads to the increase of thermal conductivity from 0.225 to 0.235 W/m. K. Wang et al. [41] prepared a high thermal conductivity form-stable PCM by blending polyethylene glycol, silica gel and β -Aluminum nitride (β -AlN) powder. The thermal conductivity of the composite PCMs increased from 0.2985 W/m.K (pure polyethylene glycol) to 0.7661 W/m.K when the amount of β -AlN was 30 wt%.

2.3. Comparison of different ways to enhance the thermal conductivity of Phase change materials

Among the methods mentioned in our review we can positively opine that using the carbon additives could be a good choice to enhance the thermal conductivity due to their low density, thermal stability and diversity. As for metal additives, they have a higher thermal conductivity enhancement, but their use is limited due to their instability during thermal cycling. However, the selection of those enhancement methods is based on the price, experimental conditions and reliability.

Conclusions and outlook

A review of experimental work to enhance the thermal conductivity of phase change materials was discussed. The actual review presents different approaches to improve the thermal conductivity, using carbon additives such as: expanded graphite, carbon fiber, graphene, etc or metal additives like copper foam, nickel foam, etc. in both categories, the thermal conductivity of phase change material composites has been enhanced. However, the selection of any additive is based on the stability of both components (PCM and additive) in the composite, the economic costs and environmental impact etc.

As a matter fact, we believe that the wettability of PCM on the additive surface is also an important parameter that should be taken by consideration before preparing PCM/additive composite. In literature, there are few studies on the wettability of high thermal conductivity enhancer (carbon additives or metals foams) by PCM. The preparation of new material with good wetting could be a novel research idea and approach.

References

1. E. Foster, M. Contestabile, J. Blazquez, B. Manzano, M. Workman, N. Shah, The unstudied barriers to widespread renewable energy deployment, Fossil fuel price responses. *Energy Policy*, 103 (2017) 258-264. <https://doi.org/10.1016/j.enpol.2016.12.050>
2. B. Rennkamp, S. Haunss, K. Wongs, A. Ortega, E. Casamadrid, Competing coalitions: The politics of renewable energy and fossil fuels in Mexico, South Africa and Thailand, *Energy Research & Social Science*, 34 (2017) 214-223. <https://doi.org/10.1016/j.erss.2017.07.012>
3. T. M. Letcher, (Ed.), Future energy: improved, sustainable and clean options for our planet, *Elsevier*. (2020). <https://doi.org/10.1515/ci.2009.31.1.22>
4. S. Wu, T. Yan, Z. Kuai, W. Pan, Thermal conductivity enhancement on phase change materials for thermal energy storage: A review, *Energy Storage Materials* (2019). [10.1016/j.ensm.2019.10.010](https://doi.org/10.1016/j.ensm.2019.10.010)
5. G. Alva, Y. Lin, G. Fang, An overview of thermal energy storage systems, *Energy*, 144 (2018) 341-378. <https://doi.org/10.1016/j.energy.2017.12.037>
6. A. M. Khudhair, M. M. Farid, A review on energy conservation in building applications with thermal storage by latent heat using phase change materials, *Energy conversion and management*, 45(2) (2004) 263-275. [https://doi.org/10.1016/S0196-8904\(03\)00131-6](https://doi.org/10.1016/S0196-8904(03)00131-6)
7. M. K. Rathod, J. Banerjee, Thermal stability of phase change materials used in latent heat energy storage systems: a review, *Renewable and sustainable energy reviews*, 18 (2013) 246-258. <https://doi.org/10.1016/j.rser.2012.10.022>
8. A. Sharma, V. V. Tyagi, C. R. Chen, D. Buddhi, Review on thermal energy storage with phase change materials and applications, *Renewable and Sustainable energy reviews*, 13(2) (2009) 318-345. <https://doi.org/10.1016/j.rser.2007.10.005>
9. M. N. A. Hawlader, M. S. Uddin, H. J. Zhu, Encapsulated phase change materials for thermal energy storage: experiments and simulation, *International Journal of Energy Research*, 26(2) (2002) 159-171. <https://doi.org/10.1002/er.773>
10. B. Zalba, J. M. Marin, L. F. Cabeza, H. Mehling, Review on thermal energy storage with phase change: materials, heat transfer analysis and applications, *Applied thermal engineering*, 23(3) (2003) 251-283. [https://doi.org/10.1016/S1359-4311\(02\)00192-8](https://doi.org/10.1016/S1359-4311(02)00192-8)
11. G. Li, Y. Hwang, R. Radermacher, H. H. Chun, Review of cold storage materials for subzero applications, *Energy*, 51 (2013) 1-17. <https://doi.org/10.1016/j.energy.2012.12.002>
12. L. Fan, J. M. Khodadadi, Thermal conductivity enhancement of phase change materials for thermal energy storage: a review, *Renewable and sustainable energy reviews*, 15(1) (2011) 24-46. <https://doi.org/10.1016/j.rser.2010.08.007>
13. A. Mills, M. Farid, J. R. Selman, S. Al-Hallaj, Thermal conductivity enhancement of phase change materials using a graphite matrix, *Applied Thermal Engineering*, 26(14-15) (2006) 1652-1661. <https://doi.org/10.1016/j.applthermaleng.2005.11.022>
14. S. Kim, L. T. Drzal, High latent heat storage and high thermal conductive phase change materials using exfoliated graphite nanoplatelets, *Solar Energy Materials and Solar Cells*, 93(1) (2009) 136-142. <https://doi.org/10.1016/j.solmat.2008.09.010>
15. T. Xu, Y. Li, J. Chen, J. Liu, Preparation and thermal energy storage properties of LiNO₃-KCl-NaNO₃/expanded graphite composite phase change material, *Solar Energy Materials and Solar Cells*, 169(2017) , 215-221. <https://doi.org/10.1016/j.solmat.2017.05.035>
16. Z. Song, Y. Deng, J. Li, H. Nian, Expanded graphite for thermal conductivity and reliability enhancement and supercooling decrease of MgCl₂·6H₂O phase change material, *Materials Research Bulletin*, 102(2018), 203-208. <https://doi.org/10.1016/j.materresbull.2018.02.024>

17. L. Zhong, X. Zhang, Y. Luan, G. Wang, Y. Feng, D. Feng, Preparation and thermal properties of porous heterogeneous composite phase change materials based on molten salts/expanded graphite, *Solar Energy*, 107 (2014) 63-73. <https://doi.org/10.1016/j.solener.2014.05.019>
18. Y. Wu, T. Wang, Hydrated salts/expanded graphite composite with high thermal conductivity as a shape-stabilized phase change material for thermal energy storage, *Energy conversion and management*, 101 (2015) 164-171. <https://doi.org/10.1016/j.enconman.2015.05.006>
19. H. Tian, W. Wang, J. Ding, X. Wei, C. Huang, Preparation of binary eutectic chloride/expanded graphite as high-temperature thermal energy storage materials, *Solar Energy Materials and Solar Cells*, 149 (2016) 187-194. <https://doi.org/10.1016/j.solmat.2015.12.038>
20. Z. Huang, X. Gao, T. Xu, Y. Fang, Z. Zhang, Thermal property measurement and heat storage analysis of LiNO₃/KCl–expanded graphite composite phase change material, *Applied energy*, 115 (2014) 265-271. <https://doi.org/10.1016/j.apenergy.2013.11.019>
21. J. Xiao, J. Huang, P. Zhu, C. Wang, X. Li, Preparation, characterization and thermal properties of binary nitrate salts/expanded graphite as composite phase change material, *Thermochimica acta*, 587 (2014) 52-58. <https://doi.org/10.1016/j.tca.2014.04.021>
22. A. Elgafy, K. Lafdi, Effect of carbon nanofiber additives on thermal behavior of phase change materials, *Carbon*, 43(15) (2005) 3067-3074. <https://doi.org/10.1016/j.carbon.2005.06.042>
23. L. Fan, J. M. Khodadadi, Thermal conductivity enhancement of phase change materials for thermal energy storage: a review, *Renewable and sustainable energy reviews*, 15(1) (2011) 24-46. <https://doi.org/10.1016/j.rser.2010.08.007>
24. B. Tian, W. Yang, L. Luo, J. Wang, K. Zhang, J. Fan, T. Xing, Synergistic enhancement of thermal conductivity for expanded graphite and carbon fiber in paraffin/EVA form-stable phase change materials, *Solar Energy*, 127 (2016) 48-55. <https://doi.org/10.1016/j.solener.2016.01.011>
25. Y. Cai, H. Ke, J. Dong, Q. Wei, J. Lin, Y. Zhao, H. Fong, Effects of nano-SiO₂ on morphology, thermal energy storage, thermal stability, and combustion properties of electrospun lauric acid/PET ultrafine composite fibers as form-stable phase change materials, *Applied Energy*, 88(6) (2011) 2106-2112. <https://doi.org/10.1016/j.apenergy.2010.12.071>
26. Z. Jiang, T. Ouyang, Y. Yang, L. Chen, X. Fan, Y. Chen, Y. Fei, Thermal conductivity enhancement of phase change materials with form-stable carbon bonded carbon fiber network, *Materials & Design*, 143 (2018) 177-184. <https://doi.org/10.1016/j.matdes.2018.01.052>
27. T. Nomura, K. Tabuchi, C. Zhu, N. Sheng, S. Wang, T. Akiyama, High thermal conductivity phase change composite with percolating carbon fiber network, *Applied energy*, 154 (2015) 678-685. <https://doi.org/10.1016/j.apenergy.2015.05.042>
28. F. Frusteri, V. Leonardi, S. Vasta, G. Restuccia, Thermal conductivity measurement of a PCM based storage system containing carbon fibers, *Applied thermal engineering*, 25(11-12) (2005) 1623-1633. <https://doi.org/10.1016/j.applthermaleng.2004.10.007>
29. T. Qian, J. Li, W. Feng, Enhanced thermal conductivity of form-stable phase change composite with single-walled carbon nanotubes for thermal energy storage, *Scientific reports*, 7 (2017) 44710. [https://doi.org/10.1038/srep44710\(2017\)](https://doi.org/10.1038/srep44710(2017))
30. X. Zhang, R. Wen, Z. Huang, C. Tang, Y. Huang, Y. Liu, Y. Xu, Enhancement of thermal conductivity by the introduction of carbon nanotubes as a filler in paraffin/expanded perlite form-stable phase-change materials, *Energy and Buildings*, 149 (2017) 463-470. <https://doi.org/10.1016/j.enbuild.2017.05.037>
31. F. Ye, Z. Ge, Y. Ding, J. Yang, Multi-walled carbon nanotubes added to Na₂CO₃/MgO composites for thermal energy storage, *Particuology* 15 (2014) 56-60. [10.1016/j.partic.2013.05.001](https://doi.org/10.1016/j.partic.2013.05.001)

32. F. Ye, Z. Ge, Y. Ding, J. Yang, Multi-walled carbon nanotubes added to Na₂CO₃/MgO composites for thermal energy storage, *Particuology*, 15 (2014) 56-60. [10.1016/j.partic.2013.05.001](https://doi.org/10.1016/j.partic.2013.05.001)
33. Y. B. Tao, C. H. Lin, Y. L. He, Preparation and thermal properties characterization of carbonate salt/carbon nanomaterial composite phase change material, *Energy conversion and management*, 97 (2015) 103-110. <https://doi.org/10.1016/j.enconman.2015.03.051>
34. H. Tian, W. Wang, J. Ding, X. Wei, M. Song, J. Yang, Thermal conductivities and characteristics of ternary eutectic chloride/expanded graphite thermal energy storage composites, *Applied energy*, 148 (2015) 87-92. <https://doi.org/10.1016/j.apenergy.2015.03.020>
35. X. Wang, Q. Guo, J. Wang, Y. Zhong, L. Wang, X. Wei, L. Liu, Thermal conductivity enhancement of form-stable phase-change composites by milling of expanded graphite, micro-capsules and polyethylene, *Renewable energy*, 60 (2013) 506-509. [10.1016/j.renene.2013.05.038](https://doi.org/10.1016/j.renene.2013.05.038)
36. M. Li, M. Chen, Z. Wu, J. Liu, Carbon nanotube grafted with polyalcohol and its influence on the thermal conductivity of phase change material, *Energy conversion and management*, 83 (2014) 325-329. <https://doi.org/10.1016/j.enconman.2014.04.002>
37. M. Xing, J. Yu, R. Wang, Experimental study on the thermal conductivity enhancement of water based nanofluids using different types of carbon nanotubes, *International Journal of Heat and Mass Transfer*, 88 (2015) 609-616. <https://doi.org/10.1016/j.ijheatmasstransfer.2015.05.005>
38. S. Azeem, M. Zain-ul-Abdein, Investigation of thermal conductivity enhancement in bakelite-graphite particulate filled polymeric composite, *International Journal of Engineering Science*, 52 (2012) 30-40. <https://doi.org/10.1016/j.ijengsci.2011.12.002>
39. J. N. Shi, M. D. Ger, Y. M. Liu, Y. C. Fan, N. T. Wen, C. K. Lin, N. W. Pu, Improving the thermal conductivity and shape-stabilization of phase change materials using nanographite additives, *Carbon*, 51(2013) 365-372. <https://doi.org/10.1016/j.carbon.2012.08.068>
40. X. Xiao, P. Zhang, M. Li, Effective thermal conductivity of open-cell metal foams impregnated with pure paraffin for latent heat storage, *International Journal of Thermal Sciences*, 81 (2014) 94-105. <https://doi.org/10.1016/j.ijthermalsci.2014.03.006>
41. W. Wang, X. Yang, Y. Fang, J. Ding, J. Yan, Enhanced thermal conductivity and thermal performance of form-stable composite phase change materials by using β -Aluminum nitride, *Applied Energy*, 86(7-8) (2009) 1196-1200. <https://doi.org/10.1016/j.apenergy.2008.10.020>
42. K. Chintakrinda, R. D. Weinstein, A. S. Fleischer, A direct comparison of three different material enhancement methods on the transient thermal response of paraffin phase change material exposed to high heat fluxes, *International Journal of Thermal Sciences*, 50(9) (2011) 1639-1647. <https://doi.org/10.1016/j.ijthermalsci.2011.04.005>
43. X. Xiao, P. Zhang, M. Li, Preparation and thermal characterization of paraffin/metal foam composite phase change material, *Applied energy*, 112 (2013) 1357-1366. <https://doi.org/10.1016/j.apenergy.2013.04.050>
44. S. Thapa, S. Chukwu, A. Khaliq, L. Weiss, Fabrication and analysis of small-scale thermal energy storage with conductivity enhancement, *Energy conversion and management*, 79 (2014) 161-170. <https://doi.org/10.1016/j.enconman.2013.12.019>
45. C. A. Bauer, R. A. Wirtz, Thermal characteristics of a compact, passive thermal energy storage device, *ASME-PUBLICATIONS-HTD*, 366 (2000) 283-290.
46. J. Jiang, Y. Zhu, A. Ma, D. Yang, F. Lu, J. Chen, D. Song, Preparation and performances of bulk porous Al foams impregnated with phase-change-materials for thermal storage, *Progress in Natural Science: Materials International*, 22(5) (2012) 440-444. <https://doi.org/10.1016/j.pnsc.2012.05.004>

47. T. X. Li, D. L. Wu, F. He, R. Z. Wang, Experimental investigation on copper foam/hydrated salt composite phase change material for thermal energy storage, *International Journal of Heat and Mass Transfer*, 115(2017) 148-157. <https://doi.org/10.1016/j.ijheatmasstransfer.2017.07.056>
48. X. Huang, Y. Lin, G. Alva, G. Fang, Thermal properties and thermal conductivity enhancement of composite phase change materials using myristyl alcohol/metal foam for solar thermal storage, *Solar Energy Materials and Solar Cells*, 170 (2017) 68-76. <https://doi.org/10.1016/j.solmat.2017.05.059>
49. H. Fauzi, H. S. Metselaar, T. M. I. Mahlia, M. Silakhori, Sodium laurate enhancements the thermal properties and thermal conductivity of eutectic fatty acid as phase change material (PCM), *Solar Energy*, 102 (2014) 333-337. <https://doi.org/10.1016/j.solener.2013.07.001>

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